

2007-2008 NCLT Professional Development Summer Institute



Professional Development Work Circle Team

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FOLLOW-UP ACTIVITIES

NEW LESSONS

NEW LESSONS

Poster Session

- Teachers created posters on a nano-lesson they implemented in their class during the year.
- Many lessons implemented were on size & scale and IMF.

Research Seminar

- Bionanotechnology: Dr. Alex Wei (Purdue)
- Size and Scale: Dr. Shawn Stevens (Michigan)

Big Ideas Workshop

- Led by Dr. Shawn Stevens (Michigan)
- Teachers formed small groups and discussed a particular big idea
- Large group discussion ensued at the conclusion

Focus Group Evaluation

- Positive feedback given by teachers

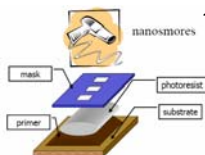
Ferrofluids

- Learning Goals**
 - Perform the steps to synthesize a ferrofluid.
 - Explain how the physical properties (appearance, viscosity) of a ferrofluid function in relation to its magnetic behavior.
 - Understand the interdisciplinary nature of ferrofluids and nanotechnology and the many current and potential future applications of ferrofluids.
 - Draw the magnetic field lines of force from a magnet and compare them to the spikes exhibited by a ferrofluid in a magnetic field.



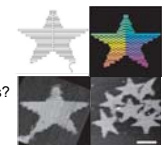
Lithography

- Learning Goals**
 - Lithographic techniques
 - Applications of lithography
 - Chemical reactions
 - Properties of light
 - Moore's Law
 - Electron beam lithography
 - Modeling of lithography



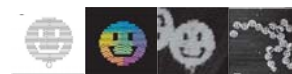
DNA Origami

- Inquiry Questions**
 - How do scientists and engineers get DNA to self-assemble into shapes?
- Learning Goals**
 - DNA can fold.
 - Each DNA base is attracted to a specific other base (A and T, C and G). These are called base pairs.
 - Components and their environment can be designed so that a patterned structure results from a self-assembly process.
 - An AFM scan allows scientists to determine if their designed DNA pattern has formed.



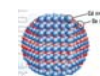
Skills

- Reading and interpreting scientific literature
- Communicating concepts to other students
- Designing representative models of science concepts
- Critiquing models of science concepts



Quantum Dots

- Inquiry Questions**
 - What is a quantum dot?
 - How are quantum dots of specific colors synthesized?
 - What is the relationship between quantum dot color and size, and the corresponding wavelength and energy of light emitted?
- Learning Goals**
 - Reaction time of the reactants controls the color of the quantum dot.
 - As the color of a quantum dot goes from red to purple, the wavelength of light emitted becomes shorter.
 - The larger the dot, the redder the light.
 - $E=hc/\lambda$. As the wavelength of light emitted decreases, the energy associated with that wavelength increases.
 - Energy level diagrams: The energy associated with an electron in the shell of quantum dot can be modeled graphically.
 - A Transmission Electron Microscope (TEM) can be used to determine the size of nanostructures.
- Skills**
 - Designing experiments to discover trends
 - Comparing simulated experimental results to actual experimental results
 - Communicating results and evidence

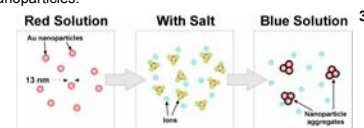


SAMPLE SUMMER INSTITUTE SCHEDULE (PURDUE)

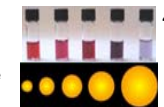
Day	8:00-11:00 AM	11:00 AM-1:00 PM	1:00-3:00 PM	3:00-5:00 PM	5:00-7:00 PM	7:00-9:00 PM	9:00-11:00 PM
Monday	Registration 8:00-11:00 AM	Registration 11:00 AM-1:00 PM	Registration 1:00-3:00 PM	Registration 3:00-5:00 PM	Registration 5:00-7:00 PM	Registration 7:00-9:00 PM	Registration 9:00-11:00 PM
Tuesday	Activity 1: Intro to Nano David Sederberg Activity 2: Synthesis of Quantum Dots David Sederberg Activity 3: Synthesis of Quantum Dots David Sederberg Activity 4: Synthesis of Quantum Dots David Sederberg Activity 5: Synthesis of Quantum Dots David Sederberg Activity 6: Synthesis of Quantum Dots David Sederberg Activity 7: Synthesis of Quantum Dots David Sederberg Activity 8: Synthesis of Quantum Dots David Sederberg Activity 9: Synthesis of Quantum Dots David Sederberg Activity 10: Synthesis of Quantum Dots David Sederberg Activity 11: Synthesis of Quantum Dots David Sederberg Activity 12: Synthesis of Quantum Dots David Sederberg Activity 13: Synthesis of Quantum Dots David Sederberg Activity 14: Synthesis of Quantum Dots David Sederberg Activity 15: Synthesis of Quantum Dots David Sederberg Activity 16: Synthesis of Quantum Dots David Sederberg Activity 17: Synthesis of Quantum Dots David Sederberg Activity 18: Synthesis of Quantum Dots David Sederberg Activity 19: Synthesis 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Biosensors

- Inquiry Opening**
 - Many people die from lead poisoning, what are some cheap easy way to detect the presence of lead or other molecules?
- Learning Goals**
 - Students will understand properties of electrolytes such as ionic versus covalent bonding.
 - Students will discover how electrolytes affect a solution of gold nanoparticles.
- Skills**
 - Designing experiments to discover trends
 - Comparing simulated experimental results to actual experimental results
 - Communicating results and evidence



- Students will learn that although a color change occurs, which usually indicates a chemical change, only a physical change is occurring rather than a chemical change.
- Students will understand spectroscopy (absorbance vs. wavelength).
- Students will understand that you see red because green light (520 nm) is absorbed.
- Students will understand that the size of the particles affects the color absorbed.
- Students will gain a deeper understanding of biosensors, what they are, what they are used for, and how they work.



CONNECTIONS

Network for Computational Nanotechnology (NCN)

- Creating simulations for activities
 - Biosensors
 - Quantum Dots

ScienceScape (Purdue)

- Outreach of nano-lessons to 7-9th grade girls

Gifted Educational Research Initiative (Purdue)

- Super Saturday: gifted middle-school students

Big Ideas Conference (Michigan and SRI)

Work Circles (NCLT)

- Nano-SLAM
- Self-Assembly
- Concept Inventory

REFERENCES

- http://www.nbc.cornell.edu/mainstreescience/nanosmores_and_photolithography.pdf
- Rothmund P.W.K. (2006). Folding DNA to create nanoscale shapes and patterns. *Nature*, 440, 297-302.
- <http://jchemed.chem.wisc.edu/HS/Journal/Issues/2004/AprilicSubscriber/JCESup.p/JCE2004p0544AW.pdf>
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- http://www2a.cdc.gov/niosh-nii/uploads01/evident_cadmium_1_1n.jpg
- <http://www.rsc.org/chemistryworld/restricted/2006/July/BiosensorsMarketBig.asp>